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The Pharma Innovation



ISSN (E): 2277-7695 ISSN (P): 2349-8242 NAAS Rating: 5.23 TPI 2022; SP-11(7): 266-269 © 2022 TPI

www.thepharmajournal.com Received: 19-04-2022 Accepted: 21-06-2022

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Synthesis of titanium dioxide nanoparticles using Spirulina platensis algae extract

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DOI: https://doi.org/10.22271/tpi.2022.v11.i7Sd.13643

Abstract

An investigation of green synthesis of titanium dioxide nanoparticles using *Spirulina platensis* algae extract was undertaken in the Department of Sericulture, Forest College and Research Institute, Mettupalayam. The nanoparticles formation, particle size, exterior morphology and shape of synthesized TiO₂ NPs were analyzed through UV-Vis absorbance spectroscopy, X-Ray diffraction and Scanning electron microscope (SEM) respectively. The results showed that the absorbance spectra of synthesized NPs were observed at 300nm. X-Ray diffraction analysis revealed that the formation of good crystalline TiO₂ NPs with anatase phase. The synthesised NPs were in spherical shape and dispersed irregularly. The green synthesised of TiO₂ NPs is a simple approach, inexpensive and eco-friendly process which having potential applications in various fields.

Keywords: TiO2 NPs, green synthesis, S. platensis, characterization

Introduction

The nanomaterials are receiving considerable attention because of their exceptional physicochemically properties over bulk counterparts. Nanotechnology is a field of science with enormous potential in medicine. Nanotechnology plays an extremely significant role in present research as it is the most proficient technology that can be used in almost all the fields such as cosmetics, pharmaceuticals, environmental health, food and feed, chemical industry, agricultural science, energy sector, drug and gene delivery, mechanics and space industry. The synthesis of nanomaterials using chemical methods is very expensive, requires complex steps and leads to the absorption of toxic chemicals on the surface that may have an adverse effect on the agricultural application and are hazardous to the environment. Therefore, the development of clean, biocompatibility, non-toxic and eco-friendly methods using microorganisms, enzymes and plant extract is gaining importance in nanotechnology.

Titanium dioxide nanoparticles are natural mineral oxide widely used in pharmaceuticals, cosmetics, food colouring and implantable biomaterials because of their physicochemical properties such as photocatalyst, antimicrobial agent and preservatives (Gao *et al.*, 2003) ^[1]. Synthesis of nanoparticles under eco-sustainable, nontoxic green conditions is of greater importance to combat the rising concerns about the toxicity of metallic nanomaterials in medical and technological applications. Plant extracts may act both as reducing agents in the synthesis of nanoparticles. The source of plant extract is known to have an impact on the characteristics of nanoparticles (Kumar and Yadav, 2009) ^[2]. Among various semiconductor nanoparticles, titanium dioxide (TiO₂) nanomaterial has been broadly recognised as semiconductors as they are a more ideal component for environment and energy applications because of their identical properties (Mohamed, 2012) ^[3].

Green synthesis of nanoparticles from plant extracts and microbes has attracted the attention of researchers in recent years. Biosynthesis of nanoparticles is considered better than chemical synthesis because of the formation of toxic chemical species which are adsorbed on the particle surface after chemical synthesis. Moreover, they are cost-effective and environmental friendly in nature due to a biological process, which makes them superior to chemical and physical synthesis (Parashar *et al.*, 2009)^[4].

The blue-green algae *Spirulina platensis* contains various minerals and 18 amino acids such as glycine, glutamine, histidine, lysine, methionine, cysteine, creatine, phenylalanine, serine, proline, tryptophan, asparagine, pyruvic acid, and pivotal vitamins like tocopherol, biotin, thiamine, niacin, riboflavin, proterozoic acid, folic acids, beta-carotene and vitamin B12, etc.

(Soliman and Mohamed, 2021). In this study, we have synthesized TiO_2 NPs exploiting aqueous extract of *S. platensis* a green reducing material and stabilising agent. These biogenically synthesized TiO_2 NPs were characterized using UV-Vis spectroscopy, XRD and Scanning electron microscopy (SEM).

Materials and Methods

Algae

The dried powder of Spirulina was collected from R.K. Algae Centre in Mandapam, Tamil Nadu, India. The confirmation of algal species was done at Botanical Survey of India, Coimbatore, Tamil Nadu.

Preparation of aqueous extraction of S. platensis

The aqueous extract of *S. platensis* was obtained by heating 10g of finely ground powder in 100 ml of deionized water at 90 °C for 45 mins, then the solution was filtered through Whatman filter paper No.1 to remove the debris. The resultant clear green coloured solution obtained was stored at 4-8°C for further study (Roy *et al.*, 2019)^[13].

Synthesis of titanium dioxide nanoparticles

UV-Vis Absorbance Spectroscopy

Spirulina mediated TiO₂ NPs were synthesised utilising 0.01 mM titanium dioxide and aqueous extract of Spirulina platensis as bio-reductant and capping agent in green synthesis. Aqueous extract of 20 mL was added to 80 mL of 0.01 M TiO₂ solution, which was maintained at room temperature for 6 hours with continuous stirring in a hotplate magnetic stirrer. A colour change confirmed the production of TiO₂ NPs (pale yellow).

UV-Vis spectroscopy, XRD and SEM were used to describe the green synthesis of *Spirulina*-mediated TiO₂ NPs. The green synthesis of *Spirulina* mediated titanium dioxide nanoparticles were confirmed by optical measurement using UV-visible spectroscopy in the wavelength ranging between 200 and 800 nm at the resolution of 1 nm (Subhapriya *et al.*, 2018)^[12].

X-Ray Diffraction

The phase development and crystallinity of *Spirulina* induced TiO₂ NPs were studied using X-Ray Diffraction, an analytical technique. Smart Lab was used to record the XRD pattern of produced *Spirulina*-mediated TiO₂ NPs and Powder X software was used to compute the lattice parameters. The Scherrer's equation was used to compute the particle size (d spacing value) of the sample (Sun *et al.*, 2000)^[5].

Scanning Electron Microscope

The size and shape of nanoparticles was measured with nanoscale precision using a scanning electron microscope (SEM) (Yedurkar *et al.*, 2016)^[6].

Results and Discussion

UV-Vis Absorbance Spectroscopy

The molecule size and band hole of integrated green synthesis of titanium dioxide nanoparticles were determined using UV– vis absorbance spectroscopy Fig. 1 displays the UV absorption spectra of produced TiO₂, which showed a prominent absorption band at 300 nm. The present finding can be correlated with Swathi *et al.* (2019)^[14] who reported that spectral image displays the absorption peaks at 350 nm for green synthesis of TiO₂ nanoparticles and Ahamad *et al.* (2022) indicated the presence of a titanium dioxide band at 380–400 nm. Similarly, Rathna *et al.* (2020) revealed the good crystalline nature of green route mediated AgNPs showing a peak at 420 nm indicating the formation of nanoparticles.



Fig 1: UV-Vis absorption spectra of TiO₂ NPs synthesised using S. platensis extract

X- Ray Diffraction analysis

The XRD pattern of green synthesised of TiO₂ nanoparticles is shown in Fig. 2. With diffraction angles of 27.79, 36.45, 41.57, 44.41, 54.65, 57.01, 63.11, 64.39, 69.22, 70.09, 76.88, 82.61, 84.47, and 88.01, it indicates the formation of good crystalline titanium dioxide with anatase phase shape. The prominent peak at 27.79 in the XRD pattern of green synthesised TiO₂ nanoparticles is only connected with the crystallographic plane of TiO₂ anatase. The final material's stoichiometry is highly dependent on the partial pressure used during the synthesis. As a result, the synthesised TiO_2 nanoparticles could exhibit a variety of stoichiometries. The peaks of graph are in good agreement with the literature report (Akarsu *et al.*, 2006)^[8]. The location of peaks was compared to literature values and the presence of titanium dioxide particles was confirmed. The average size of the particles was calculated using Debye-Scherrer's formula.



Fig 2: XRD pattern of biosynthesised TiO2 NPs using Spirulina platensis extract

Scanning Electron Microscope

Scanning electronic microscopy was used to investigate the form, size, and surface features such as morphology. Fig 3 & 3a. clearly showed an SEM images of synthesised TiO2 nanoparticles made with *S. platensis* aqueous extract which are irregularly dispersed and spherical in shape. The average size of the produced nanoparticles using SEM images were found to be in the range of 90 to 150 nm. Fig 3 denotes a 5μ m scale range, while fig.3a denotes a 4μ m scale range. Similarly

green synthesised metal nanoparticles loaded ultrasonicassisted *S. platensis* algal extract showed rougher and irregular spores developed on the surface of UASP and confirmed that the metal nanoparticles (Ag, Cr, Pb and Zn) were successfully coated on UASP material (Gunasundari *et al.*, 2017)^[9]. In line with this result of Santhosh Kumar *et al*. (2014)^[10] revealed that SEM images of *Psidium guajava* mediated TiO₂ NPs were smooth and sphericalin shape with various physical morphology, particle size and aspect ratio.



Fig 3: SEM image of synthesised TiO2 NPs from S. platensis extract

Conclusion

The TiO₂ nanoparticles were successfully synthesised with *S. platensis* aqueous extract. The synthesised TiO₂ NPs were investigated by UV-Vis, XRD and SEM which indicated the properties of TiO₂ NPs. The UV-Vis spectral result showed that nanoparticles synthesised properly, XRD result predicte that the particles were crystalline, the Sem result displayed that the sample was nearly spherical in shape. The synthesis of NPs using green approach is simple, inexpensive and eco-friendly process, which reduces the use of toxic chemicals.

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