



ISSN (E): 2277- 7695
ISSN (P): 2349-8242
NAAS Rating: 5.23
TPI 2021; SP-10(10): 918-923
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www.thepharmajournal.com

Received: 13-08-2021
Accepted: 18-09-2021

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Changes in physico-chemical parameters along vertical gradient in Lake Manasbal, Kashmir

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Abstract

The vast resources of water in form of lakes, wetlands, streams and rivers, due to their peculiar natural conditions, are prime centres of cold water fisheries and thus have a major contribution in its development. Manasbal Lake is the deepest freshwater valley lake of Kashmir located in Ganderbal district at mean latitude of 34°15' N and longitude of 74°40'E. Manasbal Lake is warm monomictic urban lake in Kashmir and remains thermally stratified for 8–9 Months. The present study on Manasbal Lake, was undertaken from December 2019 – November 2020 to determine various physico-chemical parameters like Temperature, Depth, Dissolved oxygen, pH, Chloride, Free Carbon Dioxide, Ammoniacal Nitrogen, Nitrate Nitrogen and Total Phosphorous along vertical gradient. The bottom levels of the lake was found to be oxygen deficit ($O_2 < 1$ mg/L), which is indicative of the water body's more eutrophic nature. Clinograde type of Dissolved Oxygen curve was recorded, also pH values showed a decreasing trend while Free Carbon dioxide, Chloride, Nitrate Nitrogen, Ammoniacal Nitrogen and Total Phosphate showed increasing trend from surface to bottom waters.

Keywords: Manasbal Lake, water quality, clinograde, eutrophic, vertical gradient

1. Introduction

Lakes have been defined as a body of standing water, occupying a basin or lacking continuity with sea (Forel, 1892) [9]. They are inland water bodies which are formed in rock basins of various sizes and shapes. The lakes in Kashmir are an important fishery resource and natural water reservoir for the people of the valley, since ancient time and have substantial impact on social and economic status of the local population as it provides potable water and fish, besides being used for recreation and irrigation of the agricultural land. The origin of the lakes in Kashmir is either tectonic or fluvial, as all the lakes lie on the flood plains of river Jhelum.

The lakes of Kashmir are categorized into three different types: (i) Glacial mountain lakes, (ii) Pine forest lakes, and (iii) Valley lakes based on their origin, altitudinal situation and nature of biota they contain (Zutshi *et al.*, 1972, 1980; Kaul, 1977; Trisal, 1985) [51, 52, 17, 39]. The first series of lakes (Alipather, Gangabal, Kousernag, Kishansar, Marsar, Tarsar, Nundkhol, Toulia, Vishansar, Sheshnag) are classified as upper mountain lakes, forest lakes include Nilnag and the valley lakes include Dal, Anchar, Khanpur, Manasbal, Wular, Naranbagh, Tigam, Tilwan, Pashakuri (Pandit, 2002) [27]. The high altitude lakes are fed by snow-melt, precipitation and springs, whereas lakes of lower altitudes receive water from rivers, streams, and springs. High altitude lakes like Gangabal, Sheshnag, Tarsar, Marsar, Kausarnag, etc (alt. 3000-4000 m) in Kashmir support fisheries while the relatively low altitude lakes of Kashmir (alt. 1587-1600 m) like Anchar, Dal, Nigeen, Manasbal and Wular lying in the flood plain of river Jhelum, have abundant macrophytic vegetation and support commercial fisheries. The escalating anthropogenic pressure in recent years, in and around Himalayan aquatic ecosystems including their watersheds has contributed to the mineral enrichment of these systems, leading to accelerated eutrophication. These anthropogenic effects not only degrade water quality but also have an effect on aquatic life in lakes (Kaul, 1979) [18]. Parameters such as temperature, turbidity, nutrients, hardness, alkalinity, dissolved oxygen etc are some of the important factors that determine the growth of living organisms in the water body (Smitha, 2013) [33]. Hence, water quality assessment involves the analysis of physico-chemical, biological and microbiological parameters that reflect the biotic and abiotic status of the ecosystem (Verma *et al.*, 2012) [40].

2. Material and Methods

2.1. Study area

Manasbal lake is the deepest freshwater valley lake of Kashmir, having an area of 280 ha, 25 ha of which is marshy, situated about 32 km northwest of Srinagar city at Safapur village in district Ganderbal, The lake is oblong in outline and extends in a northeast, southwest direction with a maximum length and breadth of 3.5 km and 1.5 km respectively (Abubakr & Kundangar, 2008) [1]. The Lake is positioned at mean latitude of 34°15' N and longitude of 74°40'E, at an altitude of 1583 meters above sea level and is fed mainly by groundwater and seasonally by an irrigational stream Lar kull, on the eastern side, which flows during summer season. It drains into the river Jhelum near village sumbal through a 1.6 km Nunyar Nalla. The Lake covers an actual area of about 2.81 km² (Wanganeo, 1984; Yousuf, 1988; Yousuf, 1992) [41, 48, 49]. Maximum length 3.5 km, Maximum breadth 1.25 km, Mean breadth 0.802 km, Maximum depth 13 m, Length of shoreline 10.24 km, Total volume of water 12.8 x 10⁶ m³, Catchment area 30 km² (Wanganeo, 1984) [41].

2.1.1 Sampling site

This is the central site of the Lake, located near Cage Culture, (installed by Fisheries Department), opposite Jaroka Bagh, This is the lake's deepest portion, which remains devoid of aquatic vegetation. Water samples were collected from the pelagic zone vertically at different depth intervals surface, 3 meters, 6 meters, 9 meters and bottom. (Fig.1)



Fig 1: Map representing overall view and sampling site in Manasbal Lake

2.2 Water Sampling

Water samples were collected from various depths by using a Ruttner Sampler from the sampling site, and then water was collected in one litre polyethylene bottles marked distinctly. For dissolved oxygen, water samples were collected in separate 125ml glass stoppered bottles and the fixation of samples was done at the site. Parameters like Air temperature, Water temperature, Depth, were determined at the sampling site and the detailed analysis for parameters like Dissolved Oxygen, pH, Total Phosphorous, Free CO₂, Chloride, Nitrate nitrogen (NO₃-N), Ammonical nitrogen (NH₄-N) was carried out in the laboratory by using the methods outlined in (APHA, 2012) [4].

2.3 Statistical Analysis

The results obtained were analysed with the help of the appropriate statistical methods using Microsoft Excel.

3. Results and Discussion

3.1 Air and Water temperature

The temperature plays an important role for controlling the

physico-chemical and biological parameters of water. It is considered the most important factor in the aquatic environment particularly for freshwater, (Singh and Mathur, 2005) [32]. Atmospheric temperature ranged from 3.2 °C in January to 26.1 °C in June (Fig.2). A characteristic depth wise variation in water temperature was recorded during different seasons. Water temperature ranged from 2.5 °C at bottom to 27.3 °C at the surface (Fig.3). The highest temperature was recorded in summer in the month of (July) at surface waters. This can be due to high solar radiation, clear atmosphere and high atmosphere temperature, (Swarnalatha and Narasingrao, 1998) [37]. The lowest temperature was reported in the month of (January) at the bottom due to cold low ambient temperature and shorter photoperiod (Bohra and Bhargava, 1977) [8].

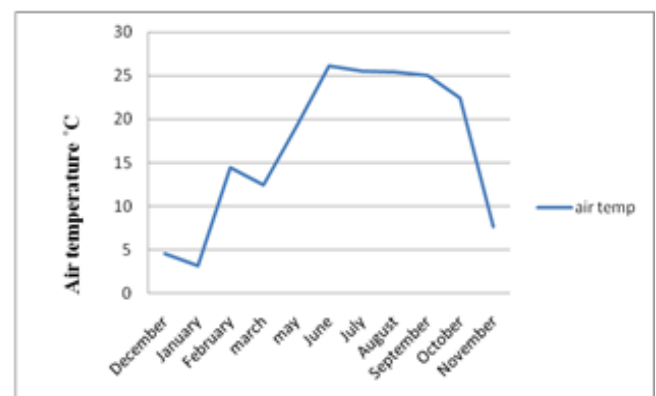


Fig 2: Monthly variation in air temperature (°C) vertically in Manasbal Lake

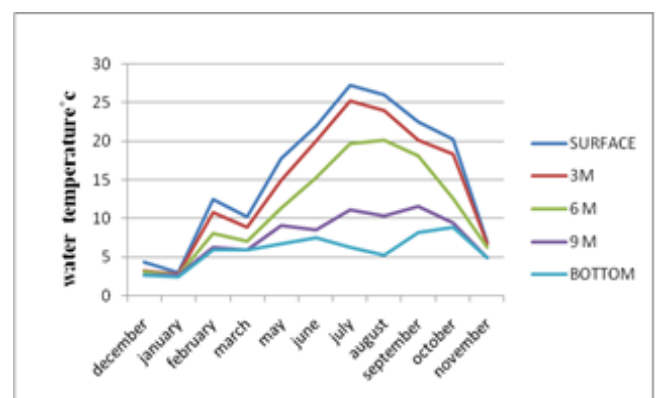


Fig 3: Monthly variation in water temperature (°C) vertically in Manasbal Lake.

3.2 Depth

The depth of a water body is one of the major physical factors which act as a controlling factor for determining the water quality. Depth of water is determined by the volume of water column in an aquatic system, which is in turn dependent on the discharge rate of inflows and the amount of precipitation received in the form of rain and other anthropogenic activities. (Bhat *et al.*, 2013 and Sushil *et al.*, 2014) [6, 36]. The depth varied from a minimum of 10.8 m in the month of July and to a maximum depth of 11.5m at site S5(M) in the month of November and December (Fig.4). Higher depth values can be accredited directly to higher precipitation rate resulting in more significant water discharge from the inlet (Kumar and Pandit, 2007) [21].

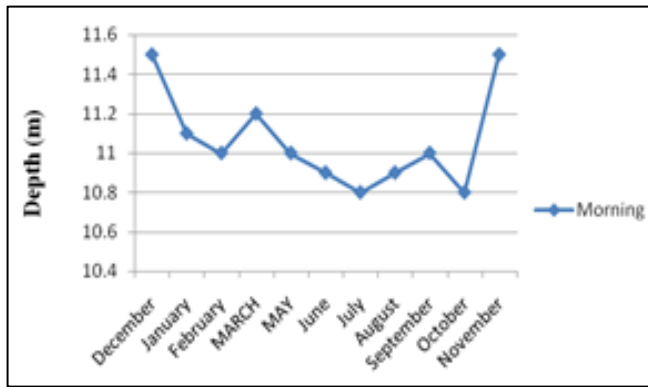


Fig 4: Monthly variation in Depth (m) at central site of Manasbal Lake

3.3 pH

pH is one of the very significant chemical characteristic of all waters, which explains certain significant biotic and abiotic ecological characteristics of aquatic systems in general. pH values showed a decreasing trend from surface to bottom during all seasons. The surface water showed a pH range from 7.4 to 8.33. The minimum pH was recorded to be 6.5 at the bottom in the month of December while a maximum pH of 8.33 was recorded at the surface in the month of June (Fig.5). pH of a water body is a diurnally variable property according to temperature variation in the system (Ojha and Mandloi, 2004) [25]. The alkaline nature of water in the epilimnion and upper layers of thermocline of water depicted the eutrophic and mesotrophic status of water (Whitemore, 1989; Kaul and Handoo, 1980) [44, 19] found that increased surface pH in water bodies is due to increased metabolic activities of autotrophs. And lower pH values in bottom layers is due liberation of acids from decomposing organic matter under low O₂ concentration result in low pH.

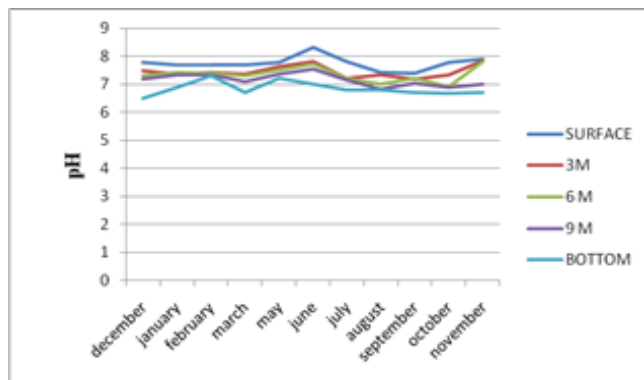


Fig 5: Monthly variation in pH vertically in Manasbal Lake.

3.4 Dissolved oxygen (D.O)

Dissolved oxygen (D.O) is a significant parameter of the water which is vital to the metabolism of every aquatic organism that possesses aerobic respiration. Dissolved Oxygen reflects the water quality status and physical and biological processes in waters as well as the metabolic balance of a lake (Laluraj *et al.*, 2002) [22]. Oxygen concentration in surface water ranged from 7.2 to 9.1 mg/l. Oxygen concentration depicted significant variations with regard to depth. Gradual decline from surface to 9m depth was recorded which finally resulted in its complete absence from June to November at 9m depth and complete absence at 12m depth depicting anoxic condition (Fig.6). Similar conditions were reported by (Jamila, 2018; Yaseen and bhat,

2021) [16, 45]. Clinograde type of curve was recorded which is characterized when productivity is high and thermal stratification occurs, oxygen depletion is likely to occur in the hypolimnion in summer and during winter stagnation under ice. This oxygen distribution is characterized by relatively higher oxygen content near the surface where photosynthesis replenishes the supply (Goldman and Horne, 1983) [11]. The oxygen deficit is caused by the high productivity of the lake, depicting meso- to eutrophic nature of lake (Pandit and Yousuf, 2002; Gunkel and Casallas, 2002) [28, 13]. The rate of oxygen depletion noticed in the bottom layers of the Manasbal lake are an indication of the more eutrophic nature of the water body. DO content is affected by fluctuations in water temperature and addition of sewage waste demanding oxygen (Koshy and Nayar, 2000) [20]. Higher DO means rate of oxygen replenishment in water is greater than O₂ consumption and this is healthy for almost all aquatic systems (Adak *et al.*, 2002) [2].

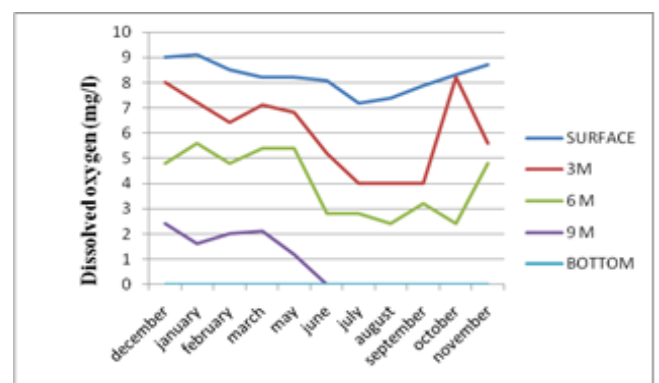


Fig 6: Monthly variation in the dissolved oxygen (mg/l) vertically in Manasbal Lake

3.5 Chloride

The concentration of chlorides can be related to purity or impurity of water (Solanki and Pandit, 2006) [34]. The chloride content varied from 5.8mg/l at surface in month of May to 32.55 mg/l at bottom in the month of December (Fig.7). The chloride content in the bottom waters was higher than surface waters which might be attributed to the presence of decomposed organic matter in the deeper waters. The high chloride concentration indicates the presence of organic matter, presumably of animal origin (Tresh *et al.*, 1944; Yousuf and Qadri, 1981) [38, 46] reveal that the chloride concentration has significantly increased over past few decades in Kashmir valley lakes and has contributed to the pollution.

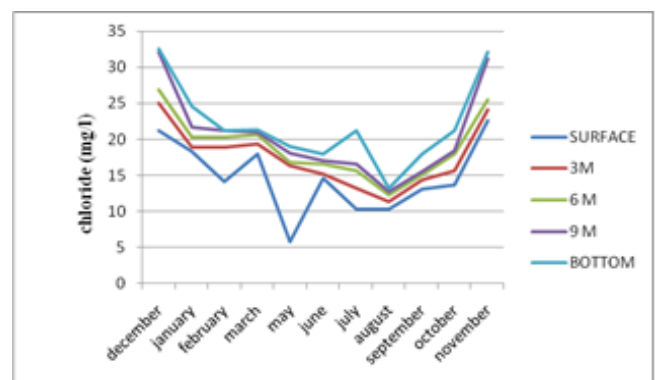


Fig 7: Monthly variation in chloride (mg/l) vertically in Manasbal Lake

3.6 Free CO₂

Free CO₂ is essential for photosynthesis and its concentration affects the phytoplankton, and its productivity. The amount of CO₂ in water usually shows an inverse relationship with oxygen (Radhika *et al.*, 2004) [30]. The surface water showed absence of free carbon dioxide in the month of June. It varied from 4.52 at surface waters to 24.64 at the bottom layer (Fig.8). A distinct vertical gradient was recorded from bottom to surface layers. In bottom layer concentration of carbon dioxide was quite high as compared to upper layers. The hypolimnetic waters were observed to contain large quantities of carbon dioxide throughout the year as a result of decomposition of organic matter, indicating the productive nature of the lake (Reche and Pace, 2002) [31].

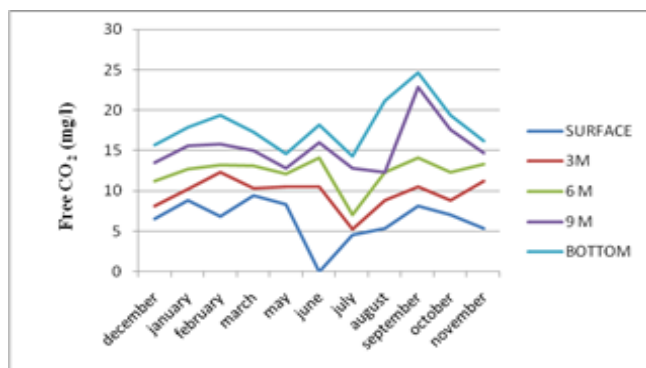


Fig 8: Monthly variation in Free CO₂ (mg/l) vertically in Manasbal Lake

3.7 Ammoniacal Nitrogen

Ammonia is basically a product of decomposition. Its concentration during the study period was generally low because of the utilization of nitrogenous compounds like NH₄-N by plants. The highest concentration of Ammoniacal Nitrogen was observed in the month of March at bottom layer (308.7 µg/l) and lowest at surface waters in July (45.24µg/l) (Fig.9). The low value of Ammoniacal Nitrogen during warmer months is attributed to the photosynthetic assimilation by autotrophs during their growth in summer (Pandit, 1999) [26]. The minimum values in the surface layers of the present lake were recorded during stagnation period as also recorded by (Yousuf, 1979) [47] for the same lake. Vertical distribution of NH₄ -N concentration in the Manasbal Lake revealed an increasing trend towards bottom. In anaerobic hypolimnion where animals are scarce, ammonium is formed at amino-acid degradation of proteins carried out by ammonifying bacteria, occurring in the water column and sediments (Gorlenko *et al.*, 1977; Howarth *et al.*, 1988; Stolp, 1996) [12, 14, 35] and thus rising ammonium-nitrogen in the bottom layers.

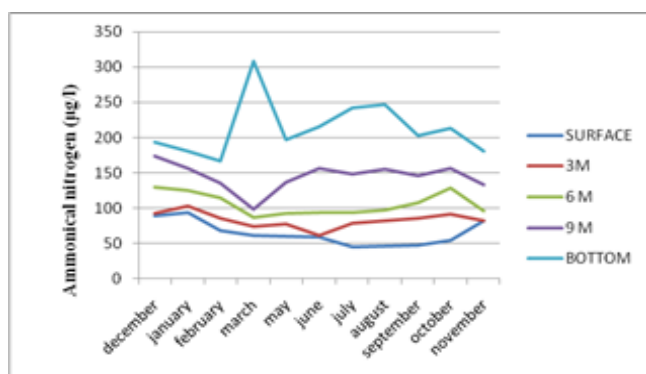


Fig 9: Monthly variation in ammonia (µg/l) vertically in Manasbal Lake

3.8 Nitrate nitrogen

Nitrate nitrogen pollution of surface waters is mainly due to discharge of industrial, municipal, domestic wastewaters and agricultural runoff including animal feedlots. Nitrogen pollution by high nitrate can cause eutrophication in surface waters (Fried, 1991; Mlitan *et al.*, 2015) [10, 24]. Nitrate is the common form of inorganic nitrogen entering the freshwater from the precipitation, ground water and draining basin, mostly occurring in low concentration under natural conditions (Wetzel., 1983) [43]. During the present study Nitrate nitrogen ranged from a minimum 191 µg/l at surface waters in the month of August to max of 393.8 µg/l at bottom layer in the month of January (Fig.10) According to (Hutchinson, 1957) [15] a well marked maximum of NO₂ is to be expected in the hypolimnion at a place where the O₂ concentration decreases most rapidly. (Anigri, 1972 and Maulood *et al.*, 1978) [3, 23] also found an increase in the NO₂ content towards bottom.

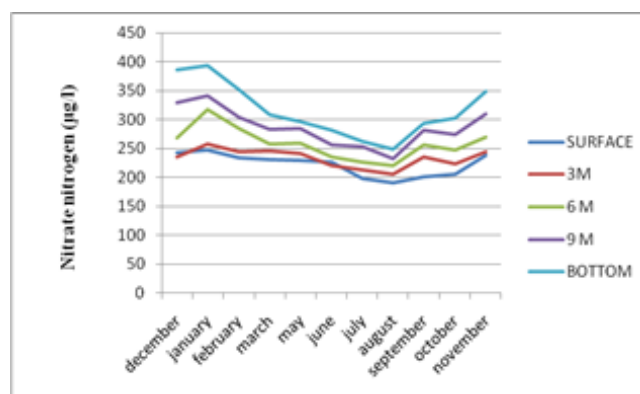


Fig 10: Monthly variation in Nitrate content (µg/l) vertically in Manasbal Lake

3.9 Total phosphorous

Phosphorus is available in water in different forms viz., orthophosphates, condensed phosphates, and organically bound phosphates (Bandela *et al.*, 1999) [5] and is considered as a critical limiting nutrient of fresh water systems (Rabalais, 2002) [29]. Presence of Phosphorous in excess, more than 30 µg/l-1 in water bodies is regarded as a major nutrient triggering eutrophication (Welch, 1980) [42]. In the present study phosphorus ranged from a minimum of 143.4 in the month of January at 3 meters to a maximum of 398.1µg/l in the month of July at bottom layer (Fig.11). No definite pattern of fluctuation was recorded in the vertical distribution in the Manasbal Lake and large variations were recorded between different layers. Constant addition of even low levels of nitrogen and phosphorous to an aquatic environment could greatly stimulate algal growth (Zimba *et al.*, 2001) [50]. Due to the anoxic conditions in hypolimnion during summer, hypolimnetic phosphorus concentration in the lake reached extremely high values due to remobilization of P from sediments (Bluszcz *et al.*, 2008) [7], same was recorded in present study in Manasbal Lake, High values of total phosphorus were recorded in summer.

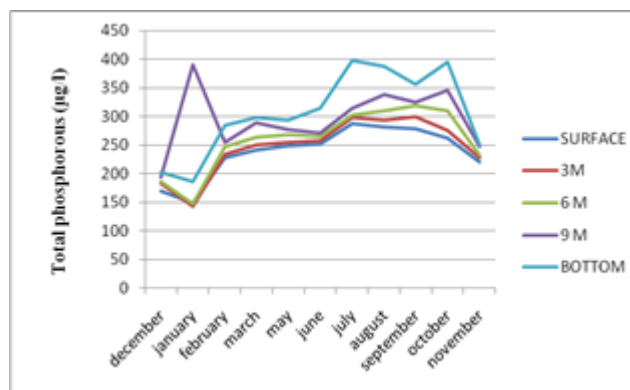


Fig 11: Monthly variation in total phosphorous content ($\mu\text{g/l}$) vertically in Manasbal Lake

Table 1: The minimum and maximum values of Physico chemical parameters at central site (vertically) of Lake Manasbal

Parameters	Surface		3 Meter		6 Meter		9 METER		BOTTOM	
	MIN	MAX	MIN	MAX	MIN	MAX	MIN	MAX	MIN	MAX
Water temperature $^{\circ}\text{C}$	3.0	27.3	2.8	25.2	2.7	20.1	2.7	11.6	2.5	8.9
pH	7.4	8.3	7.2	7.9	6.9	7.8	6.8	7.6	6.6	7.3
Dissolved oxygen (mg/l)	7.2	9.1	4.0	8.2	2.4	5.6	0.0	2.4	0.0	0.0
Chloride (mg/l)	5.8	22.7	11.3	25.0	12.3	26.9	12.7	32.1	13.2	32.6
Free CO_2 (mg/l)	4.5	9.4	5.2	12.3	7.0	14.1	12.3	22.9	14.3	24.6
Ammoniacal-Nitrogen ($\mu\text{g/l}$)	45.2	93.4	61.3	103.5	86.4	129.3	98.7	173.9	166.9	308.7
Nitrate-Nitrogen ($\mu\text{g/l}$)	191.0	247.3	206.1	257.4	221.1	316.8	232.4	341.2	249.1	393.7
Total phosphate ($\mu\text{g/l}$)	149.5	287.5	143.4	299.5	147.4	319.0	193.1	390.7	185.4	398.3

4. Conclusions

Pollution in water bodies is mainly due to addition of major plant nutrients particularly Nitrogen and Phosphorus, derived from human wastes, fertilizers, detergents, agricultural runoff etc. at an accelerated rate. The nutrients are chiefly responsible for an increase in organic production and the overall deterioration of water quality. Lake revealed a higher nutrient concentration in the deeper layers than surface layers. The lake is receiving various sources of pollution as agricultural runoff, domestic effluents and municipal wastes; this has altered the surface sediment composition, contributing to eutrophication thereby reducing the aesthetics of the lake. Agriculture land use in the catchment area and excessive use of fertilizers in the nearby Mughal gardens should be checked.

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