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Benzohydrazides: As potential bio-active agents

Deepa Kumari and Himangini Bansal

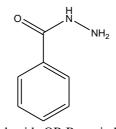
Abstract

Benzohydrazides are have gained great importance due to their diverse biological properties including anti-bacterial, anti-fungal, anti-convulsant, anti-cancer and anti-tubercular activities. This review article is comprised of information regarding all distinct benzohydrazide derivatives which were synthesised by various authors.

Keywords: benzohydrazide, MIC, tautomerisation, etc

Introduction

Hydrazines and their derivatives constitute an important class of compounds that has found wide utility in organic synthesis, while hydrazines have traditionally been employed as reagents for the derivatization and characterization of carbonyl compounds ^[1-3]. Recently hydrazides-hydrazones have gained great important due to their diverse biological properties including anti-bacterial, anti-fungal, antiinflammatory, anti-malarial and anti-tubercular activities ^[4-9]. Benzohydrazide or Benzoic-hydrazide (C₇H₈N₂O) is a heterocyclic moiety with diverse actions. It can be taken as potential pharmacophore or as a lead compound. This review is all about the previously synthesised derivatives of benzohydrazide by various authors.



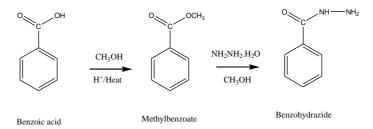
Benzohydrazide OR Benzoic-hydrazide.

Molecular weight = 136.154 g/mol, Melting point 115 °C.

General procedure for synthesis of Benzohydrazide

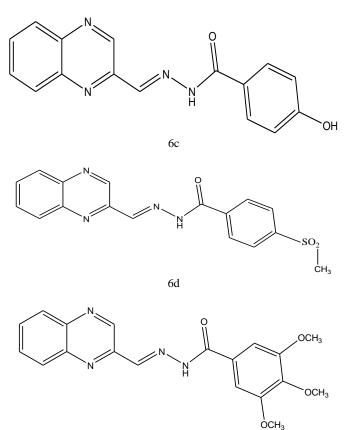
(a) **Conventional Method:** The mixture of Methyl benzoate (1.35 mL, 0.01mol) and hydrazine hydrate (0.58 mL, 0.012 mol) was taken in a flat bottomed flask and refluxed for 2 h $^{[10, 11]}$. The reaction mixture was cooled at room temperature, white precipitate was obtained. It was filtered and washed thoroughly with water.

(b) Microwave Method: The mixture of methyl benzoate (1.35 mL, 0.01 mol) and hydrazine hydrate (0.583 mL, 0.012 mol) was taken in a 100 mL beaker and was refluxed at 350 W for 2 min, then 1 mL of ethanol was added and the reaction mixture was subjected to microwave irradiation for one more minute at 500 W ^[12-14]. The resulting white precipitate was washed thoroughly with water and dried. It was further recrystallized from ethanol.



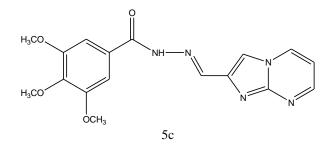
Biological profile of benzohydrazides Anti-Bacterial Activity

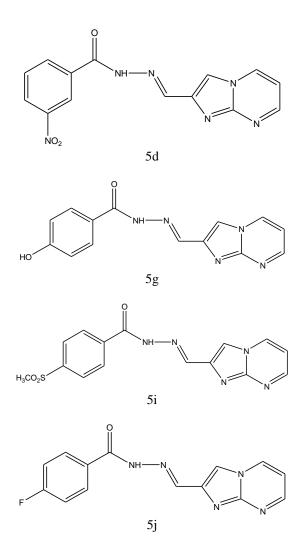
Synthesis of novel quinaxaline-benzohydrazides (6a-m) from condensation of quinoxaline-2-carboxaldehyde 4 with various benzohydrazides (5a-m) in ethanol at reflux temperature. All the derivatives were characterized by 1 H NMR, IR and mass spectroscopic analysis. The synthesized quinoxalinebenzohydrazides (6a-m) were screened for antibacterial activity. Most of the compounds shown significant antibacterial activity. Compounds 6c, 6d& 6h exhibited excellent activity as compared other synthesised compounds [15]



6h

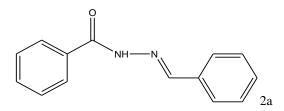
Synthesis of new imidazo [1, 2-a]pyrimidin-2-yl)methylene) benzohydrazides (5a-j) from 2-aminopyrimidine as starting material. All the synthesised compounds (5a–j) were screened for anti-bacterial activity. The Compounds 5c, 5d, 5g, 5i & 5j exhibited excellent activity as compared other synthesised compounds ^[16].



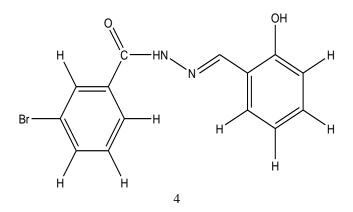


Anti-cancer activity

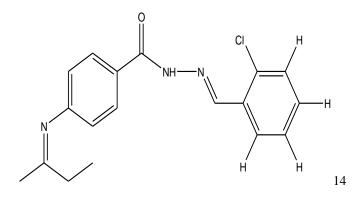
Synthesis of novel N1 [(Substituted Phenyl) Benzylidene] Benzohydrazides and were characterized by IR, 1H NMR, 13C NMR and mass spectra. The synthesized compounds (2a-2e) have been screened for their in-vitro cytotoxicity against human lung carcinoma cell lines (A-549) by MTT assay. All the compounds showed moderate to significant inhibitory activities. The synthesized compound (2a) analogueis having significant anti lung cancer activity ^[17].



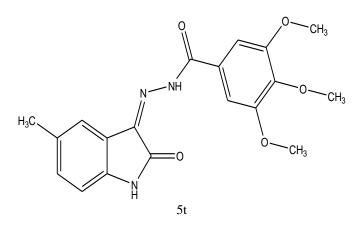
Synthesis of 2/3-bromo-N'-(substituted benzylidene/3 phenylallylidene) benzohydrazides. The synthesised compounds (1-22) were evaluated for their *in vitro* anticancer potential against human colon(HCT 116) cancer cell line. The compound 4(IC₅₀ = 1.88± 0.03 µM) was found to be the most potent anticancer agent ^[18].



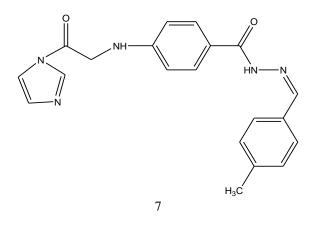
Synthesis of N'-(substituted)-4-(butan-2 lideneamino) benzohydrazides. The synthesised compounds were evaluated for their in vitro anticancer potential against human colorectal cancer cell line. The compound 14 (IC_{50} = 37.71µM) was found to be a most potent anticancer agent ^[19].



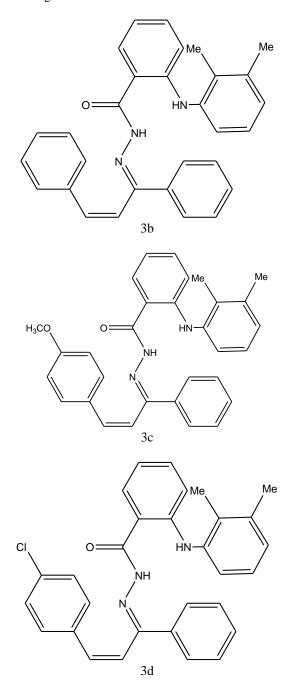
Synthesis of N'-[2-oxo1, 2 dihydro-3 H-indol-3-ylidene] benzohydrazides. The synthesised compounds (5a-5y) were evaluated for their cytotoxic activity on human cervix carcinoma grown on a plate (HeLa cells), T-cells and on another cancer cell line L1210 cells. These Benzohydrazide S show potent anti-cancer activities and are effective at much lower concentrations (nanomoles compared to micromoles) for most drugs. The compound 5t was found to be a most potent anticancer agent with an IC₅₀ value of 660 nM ^[20].



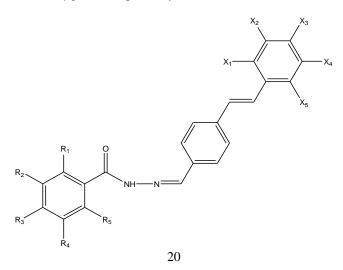
Synthesis of N '-substituted benzylidene/2hydroxynaphthalen-1-ylmethylene/3-phenylallylidene/5oxopentylidene- 4 - (2-oxo-2- (4H-1, 2, 4-triazol-4-yl) ethylamino) benzohydrazides. The synthesized compounds (1-17) were evaluated for their anticancer potentials against human colorectal cancer (HCT116) cancer cell line. The Compound 7 (IC₅₀ = 14.90 μ M) was found to be the most potent anticancer agent ^[21].



Synthesis of Nonsteroidal Anti-inflammatory Drug-based N-Allylidene Benzohydrazides. The synthesized compounds (3a-3e) were evaluated for their anticancer potentials against human colon cancer (HCT 15) cell line. The Compound 3b, 3c, 3d (IC₅₀= 13-15 μ g/ml) were found to be the most potent anticancer agent ^[22].

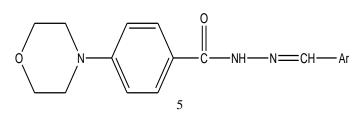


Synthesis of N'-(4-((substituted imino) methyl) benzylidene)substituted benzohydrazides. The synthesised compounds (1-48) were evaluated for their anticancer potential. Compound 20 was found to be the most effective anticancer agent against both HCT116 and MCF7 cancer cell lines, with IC₅₀ values of 19 and 18 μ g/cm3, respectively ^[23].

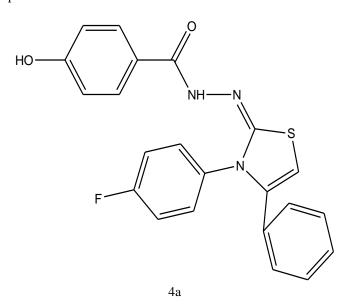


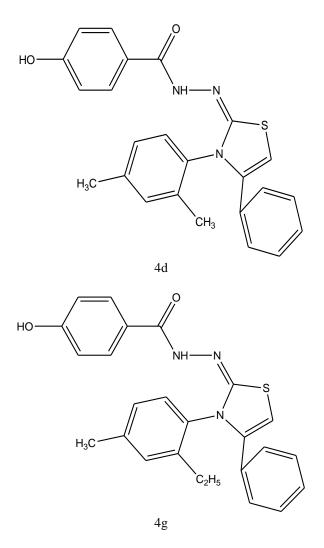
Anti-mycobacterial activity

Synthesis of Novel 4-(morpholin-4-yl)-N '-(arylidene) benzohydrazides. The synthesized compounds (5a-5j) were evaluated for their antimycobacterial activity against *Mycobacterium tuberculosis* H37Rv and compound 5c & 5d were found to be the most potent antimycobacterial agent ^[24].



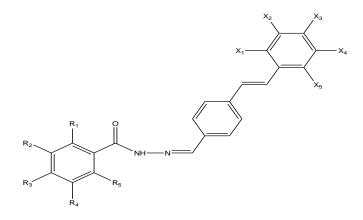
Synthesis of new p- hydroxybenzohydrazide. The synthesized compounds (4.a-4.h) were assayed *in vitro* for their antimycobacterial activity against *M. tuberculosis* H37Rv.Compounds that (4.a), (4.d) and (4.g) were found as potent inhibitor of H37RV ^[25].





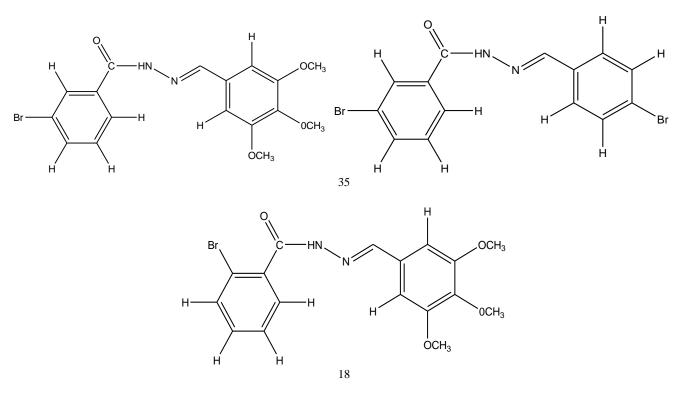
Anti-viral activity

Synthesis of N'-(4-((substituted imino) methyl) benzylidene)substituted benzohydrazides. The synthesised compounds (1-48) were evaluated for their antiviral activity against HIV-2 strain ROD. Compound 29 (IC₅₀ C 1 µg/cm3) was more potent than the standard drug nevirapine (IC₅₀ C 4 µg/cm3) and that 39 (IC₅₀ C 4 µg/cm3) was equipotent ^[26].

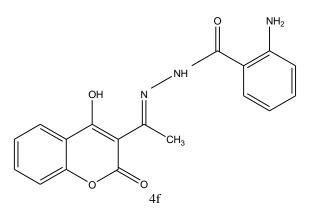


Anti-microbial activity

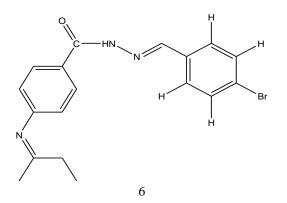
Synythesis of 2/3-bromo-N'-(substituted benzylidene/3phenylallylidene) benzohydrazides. The synthesised compounds (1-22) were evaluated for their *in vitro* antimicrobial potential against five micro-organisms. The compounds 3, 15 and 18 (pMIC_{am} = 1.62 μ M/ml) were found to be most potent anti-microbial agents ^[27].



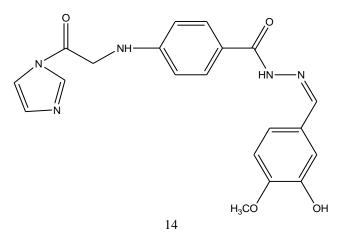
Synthesis of 2-substituted-N'-(1-(4-hydroxycumarinyl) ethylidene) benzohydrazides by condensation of 2substituted benzohydrazide with 3-acetyl-4-hydroxycumarine. The synthesised compounds (4a-4f) were evaluated against antimicrobial agent. Compound 4f was found to be a most potent antimicrobial agent ^[28].



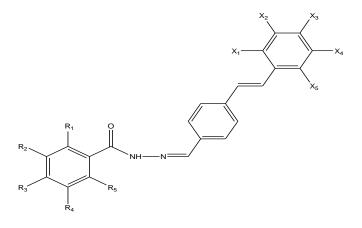
Synthesis of N'-(substituted)-4-(butan-2-lideneamino) benzohydrazides. The synthesised compounds (1-21) were evaluated against *in vitro* antimicrobial potential. The compound 6 (pMIC_{ca} = 2.07 μ M/ml) was found to be a most potent anti-fungal agent ^[29].



Synthesis of N'-substituted benzylidene/2hydroxynaphthalen-1-ylmethylene/3-phenylallylidene/5oxopentylidene-4-(2-oxo-2-(4H-1, 2, 4-triazol-4-yl) ethylamino) benzohydrazides. The synthesized compounds (1-17) were evaluated for their anti-microbial potentials & all compounds were found to be more potent against *A. niger* than other bacterial and fungal strains tested. Compound 14 (pMICan = 2.10 μ M/ml) having antifungal activity comparable to the standard drug fluconazole was found to be the most potent antifungal agent ^[30].

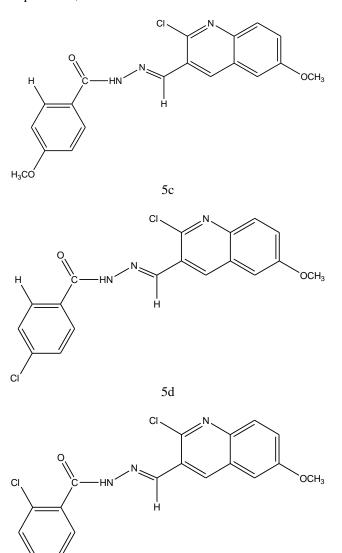


Synthesis of N'-(4-((substituted imino) methyl) benzylidene)substituted benzohydrazides. The synthesised compounds (1-48) were evaluated for their antimicrobial potential. The compound N'-[4-[(4-chlorophenylimino) methyl] benzylidene] -3-nitrobenzohydrazide (25) and N'-[4-[(2chlorophenylimino) methyl] benzylidene]-4nitrobenzohydrazide (27) (pMICam = 1.51) were the most potent antimicrobial agents ^[31].

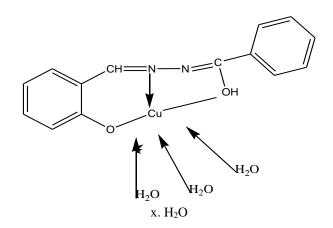


Anti-bacterial activity

Synthesis of N' -[(2-chloro-6-methoxyquinolin-3-yl) methylidene]-substituted benzohydrazide by the treatment of 2-chloro-6-methoxy-3-quinolinecarbaldehyde with the substituted benzohydrazides. The structures of the synthesized compounds have been characterized by using IR and 1H NMR spectroscopy. These compounds were screened for their antibacterial as well as antifungal activity. Compounds show greater antibacterial activity as compare to antifungal activity. However, among the series of 5a-5e synthesised compounds5c,5d & 5e were most active [32].



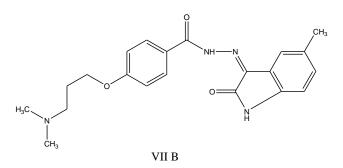
A. Padmaja et al. conducted a study on Structural Aspects and Biological (E)-N'[2-Hydroxybenzylidene] Activity of Complexes. benzohydrazide Metal and its The spectroanalytical studies of VO(II), Mn(II),Co(II),Ni(II) and Cu(II) complexes with the title compound (E)-N'-[2-Hydroxybenzylidene] benzohydrazide, as wellas antifungal and antibacterial activity of Cu(II)-HBBH complex were carried out. The complexes were characterizedby elemental analysis, IR, NMR, mass, ESR and TGA & DTA. The mode of bonding in these complexes has been suggested on the basis of analytical and spectroscopic data. The ligand coordinates to the central metal ion through oxygen of enolic form of carboxyhydrazone, phenolic oxygen and azomethine nitrogen atoms in VO(II), Mn(II), and Cu(II) complexes through the dissociation of two protons. While in Co(II) and Ni(II) complexes the dissociation of only phenolic proton is observed and hence the donor sites are oxygen of phenolic group, amide carbonyl oxygen and imine nitrogen. The Cu(II) complex exhibits lethal effect on gram negative bacteria Escherichia coli and fungal strain Candida albicans. It has also been observed that concentration of compounds play a vital role in the degree of inhibition [33].



Tentative structure of Cu (II)-HBBH Complex.

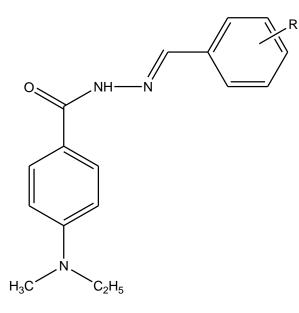
Anti-convulsant activity

Synthesis of novel NN-dialkylaminoalkoxy-2-oxoindole-3ylidene benzohydrazide derivatives. Among the synthesised (VIIa-VIIj) compounds, VIIb (R-CH3) has protected more than 80% of the mice against MES and PTZ induced convulsions. Compounds having electron donating groups at C-5 of isatin nucleus seemed to be necessary in providing higher anticonvulsant activity. Other substituents at various positions are next in the order of activity. It is evident from the anticonvulsant activity that electron donating groups seemed to be necessary moieties in showing higher anticonvulsant activity and *in vitro* docking studies have confirmed the same ^[34].

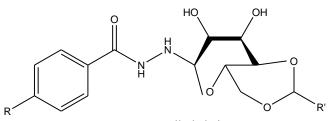


Miscellaneous activities performed with benzohydrazides

Synthesis of new N-substituted 4-(ethyl methyl amino) benzohydrazide derivatives (1-8) was synthesized using an efficient synthesis method. Reaction of these compounds with thiosalicylic acid give a new series of 1, 3-benzothiazin-4-one derivatives (9-16). The spectral (IR,1H NMR,13C NMR) and elemental analysis confirmed the structure of synthesized compounds. Their chemical transformation was studied ^[35].



Synthesis & design of long alkyl chain substituted benzohydrazide N-glycosylamines or sugar-benzohydrazides: low molecular weight organogelators. The synthesised (8-19) derivatives can form stable gels in organic solvents through Hydrogen bonding &Van Der Walls interaction ^[36].



R, R' = Long alkyl chains.

Synthesis of 4-(1H-Azol-1-ylmethyl) benzohydrazides and their Acyclic and Heterocyclic derivatives from methyl 4-(bromomethyl) benzoate, azoles, and hydrazine hydrate. Reactions of 4-(1H-imidazol-1-ylmethyl) benzohydrazide with carbonyl compounds gave hydrazones whose tautomerism was studied. From the hydrazides, 1, 3, 4-oxadiazoles, 1, 2, 4-triazole-5-thione, and N-benzoyl-N-alkyl (aryl) sulfonyl hydrazones were synthesized ^[37].

Conclusion

Benzohydrazides are selected as the main target molecules for this review as bioactive heterocyclic compounds because of their distinct biological and clinical applications. Due to its distinct biological activity, researchers synthesized variety of benzohydrazide derivatives and screened them for their various biological activities viz. anticonvulsant, antimicrobial, antimycobacterial, anticancer, antiviral, antibacterial, other miscellaneous activities. These observations based on the present review have been guiding for the development of new benzohydrazides that possess varied biological activities.

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