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Sitagliptin: A literature review on analytical and bioanalytical methods

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Abstract

This review work is a collective data of previously published methods related to the analysis of Sitagliptin either in alone or in combination with other drugs. Many spectroscopic methods like derivative techniques, chromogenic techniques were used for. Newly developed as well as improved chromatographic method also available by using biological fluids and pharmaceutical formulations. Apart from these two techniques few LC-MS/MS and HPTLC methods also available. Now in this present analytical research world quality by design or design by expert technique is used to get improved method for method validation. This concise review work can guide an analyst to choose most appropriate method for a best analytical method development and validation.

Keywords: Chromogenic, LC-ms/ms, HPTLC, Sitagliptin

1. Introduction

As pharmaceuticals developing every day, a revolution found in human health. These pharmaceuticals can show best activity if these are free from impurities and pure. At regular intervals various chemical and instrumental methods were developed to make drugs free from impurities. Impurities may develop in any stage, starting from manufacturing of bulk drug to packaging of finished product and further up to storage (degradation). Transportation and storage is the two stages where impurities may occur frequently. Hence in these condition impurities must be detected and quantitated. For detection and quantification analytical instrumentation and methods plays an important role ^[1].

For therapeutic process monitoring intermediate pharmaceutical analysis becomes an important tool as it includes different stages like testing of bulk drugs, intermediate products, drug formulations, degradation products, chemical stability of drugs and toxic contents of a drug materials. Now a day's poly pharmacy is a most worthy therapy for many diabetic patients. So for improving the poly pharmacy therapy quality control testing of combined formulations and assay of biological samples are important.

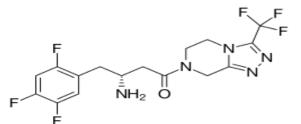
Defects in insulin secretion, insulin action, or both create diabetes which is characterized by hyperglycemia. Classification of Diabetes mellitus was widely accepted as IDDM or Type 1, and NIDDM or Type 2 which was published by WHO in 1980^[2]. DPP-4 inhibitors are the latest drugs which work by blocking the action of DPP-4, an enzyme which destroys the hormone in cretin which help the body produce more insulin only when it is needed and reduce the amount of glucose being produced by the liver when it is not needed ^[3]. The change in glucagon correlates linearly with improvement in glucose tolerance. Since these drugs improve insulin secretion in response to an increase in blood glucose, it seems appropriate to pair them with drugs that have a different mechanism of action, such as insulin sensitizers or Metformin ^[4]. During short-term clinical trials, no increased risk of acute pancreatiis has been observed with Sitagliptin, Vildagliptin, Saxagliptin, Alogliptin, and Linagliptin ^[5]. Linagliptin (Trajenta) is still included in black triangle scheme, while Sitagliptin (Januvia), Saxagliptin (Onglyza) and Vildagliptin (Galvus) were removed from the black triangle list in 2012 ^[6]. DPPIV inhibitors (Gliptins) include Saxagliptin, Linagliptin, Alogliptin, Sitagliptin, and Vildagliptin. Detail about the gliptin derivatives given in table no.1

Drug	Structure	IUPAC Name	Molecular weight	Solubility
Teneligliptin		{(2s,4s)-4-[4-(3-Methyl-1-phenyl-1-H pyrazole-5-yl) piperazin-1-yl] pyrrolidin-2- yl} (1,3-thiazolidin-3-yl) methanone hemipentahydrobromide hydrate	426.58 g/mol	Soluble in DMSO, Methanol, Water
Linagliptin		8-[(3R)-3-aminopiperidin-1-yl]-7-(but-2-yn- 1-yl)-3-methyl-1-[(4-methylquinazolin-2-yl) methyl]-2, 3, 6, 7-tetrahydro-1H-purine-2,6- dione	472.54 g/mol	It is very slightly soluble in water, IPA, Acetone, soluble in methanol, sparingly soluble in ethanol
Alogliptin		2-((6-((3R)-3-aminopiperidin-1-yl)-3- methyl-2,4-dioxo-3,4-dihydropyrimidin- 1(2H)-yl)methyl) benzonitrile.	339.39 g/mol	Soluble in Methanol; Insoluble in Acetonitrile
Sitagliptin	F F F F F	(3R)-3-amino-1-[3-(trifluoromethyl)-5,6- dihydro[1,2,4]triazolo[4,3- a]pyrazin-7(8H)- yl]-4-(2,4,5-trifluorophenyl)butan-1-one phos-phate hydrate	407.314 g/mol	It is soluble in water and N,N-dimethyl formamide; slightly soluble in methanol; very slightly soluble in ethanol, acetone, and acetonitrile; and insoluble in isopropanol and isopropyl acetate
Vildagliptin		(2S)-1-{2-[(3-hydroxy-1- adamantyl)amino]acetyl}pyrrolidine-2- carbonitrile	303.399 g/mol	Slightly soluble in DMSO, Methanol
Saxagliptin	OH NH2	(1S, 3S, 5S)-2-[(2S)-2-Amino-2-(3 hydroxytricyclo [3.3.1.13, 7] dec-1-yl) acetyl]- 2-azabicyclo [3.1.0] hexane-3- carbonitrile	315.41 g/mol	It is sparingly soluble in water, slightly soluble in ethyl acetate, and soluble in methanol, ethanol, isopropyl alcohol, acetonitrile, acetone

Sitagliptin

From all these gliptin derivatives in this present journal about sitagliptin is discussed briefly.

Sitagliptin phosphate monohydrate (SPM) chemically, (3R)-3-amino-1-[3-(trifluoromethyl)-5, 6-dihydro [1, 2, 4] triazole [4,3-a] pyrazin-7 (8H)-yl]-4-(2,4,5-trifluorophenyl) butan-1one phosphate hydrate (Fig. 1) is oral hypoglycemic drug belongs to dipeptidyl peptidase4(DPP4) inhibitor class ^[7]. DPP-4 inhibitors represent a new therapeutic approach to the treatment of type 2 diabetes that functions to stimulate glucose-dependent insulin release and reduce glucagons levels by the inhibition of the inactivation process of incretins, particularly glucagon-like peptide- 1 (GLP-1) and gastric inhibitory polypeptide (GIP), thereby improving glycemic control. Sitagliptin was approved by the U.S.



(7-[(3R)-3-amino-1-oxo-4-(2, 4,5trifluorophenyl) butyl] 5,6,7,8 tetrahydro-3-(trifluoromethyl)-1,2,4-triazolo [4,3-a] pyrazine phosphate)

Fig I: Chemical structure and IUPAC name of Sitagliptin

Food and Drug Administration (FDA) on October 17, 2006, its bioavailability 87%, protein binding 38%, metabolism via hepatic, biological half-life 8 to 14 h and root of excretion through renal. Several analytical methods based on UV, RP-HPLC, LC-MS/MS was reported for the pharmacokinetic determination of Sitagliptin phosphate in plasma and urine of

humans, rats and dogs.

This review paper focuses the analytical procedure available for the estimation of sitagliptin i.e. electrochemical methods, UV/VIS- spectrophotometric methods, HPLC/LC-MS, GC-MS, CE/CE-MS. The details about the previous studies are discussed in Table no. II, III and IV.

S. No	Stationary Phase (Column)	Mobile Phase (with ratio)	pН	Wavelength	Flow rate	Reference
		SITAGLIPTIN with METFORMIN		8		
1	C18 (100 x 4.6 mm, 5 µm)	Mixture of Methanol: Acetonitrile: Phosphate buffer (20:35:45 %, $v/v/v$)		254 nm	1ml/min	8
2	CN (250 x 4.6 mm, 5 µm)	Mixture of OPA buffer: Acetonitrile (80:20 %, v/v) 205 nr		205 nm	1ml/min	9
3	ODS (250 x 4.6 mm, 5 µm)	Mixture of Ammonium di-hydrogen phosphate buffer: Acetonitrile (74:26 % v/v)	4.3	246 nm	1 ml/min	10
4	C18 (250 x 4.6 mm, 5 µm)	Mixture of Water: Methanol (60:40 % v/v)		258 nm	1ml/min	11
5	C18 (250 x 4.6 mm, 5 µm)	Mixture of Phosphate buffer: Methanol: Acetonitrile (50:30:20 %, $v/v/v$)	4	253 nm	0.8 ml/min	12
6	C18 (100 x 4.6 mm, 5 µm)	Mixture of Potassium dihydrogen orthophosphate buffer: Methanol (50:50 % v/v)	8.5	215 nm	1 ml/min	13
7	C18 (250 x 4.6 mm, 5 µm)	Mixture of 0.02M Potassium dihydrogen phosphate buffer: Acetonitrile (60:40 % v/v)	4	252 nm	1 ml/min	14
8	C18 (250 x 4.6 mm, 5 µm)	Mixture of Methanol: Phosphate buffer (60:40 % v/v)		258 nm	1ml/min	15
9	C18 (250 x 4.6 mm, 5 µm)	Mixture of 0.01M dipotassium phosphate buffer: Acetonitrile (70:30 % v/v)	7	223 nm	1 ml/min	16
10	C8 (250 x 4.6 mm, 5 µm)	Mixture of Methanol: Water (45:55 % v/v)	3	267 nm	1ml/min	17
	· · ·	SITAGLIPTIN with SIMVASTATIN				
11	C8 (250 x 4.6 mm, 5 µm)	Mixture of Methanol: Water (70:30% v/v)	3	253 nm	1ml/min	18
12	C18 (250 x 4.6 mm, 5 µm)	Mixture of Water: Acetonitrile (30:70 % v/v)	8	236 nm	1ml/min	19
13	C18 (150 x 4.6 mm, 3.5 µm)	Mixture of phosphate buffer: Acetonitrile (30:70 % v/v)	3.5	254 nm	1ml/min	20
14	C18 (250 x 4.6 mm, 5 µm)	Mixture of Acetonitrile: Methanol: 10 mM Phosphate buffer (65:25:10 %, v/v/v)	4	250 nm	1.2 ml/min	21
15	C18 (250 x 4.6 mm, 5 µm)	Mixture of 10 mM Phosphate buffer: Acetonitrile: Methanol (45:35:20 % v/v)	6.3	255 nm	1 ml/min	22
16	C8 (250 x 4.6 mm, 5 µm)	Mixture of Methanol: Water (25:75 % v/v)	2.9	266 nm	1ml/min	23
		SITAGLIPTIN with GLICLAZIDE				
17	C8 (250 x 4.6 mm, 5 µm)	Mixture of Water: Acetonitrile (40:60 % v/v)		253 nm	1ml/min	24
		SITAGLIPTIN AS SINGLE FORMULATION				
18	C18 (250 x 4.6 mm, 5µm)	Methanol		248 nm	1ml/min	25
19	C18 (150 x 4.6 mm, 5 µm)	Mixture of 0.01M Potassium dihydrogen orthophosphate buffer: Methanol (50:50 % v/v)	2.5	267 nm	0.7 ml/min	26
20	C18 (250 x 4.6 mm, 5 µm)	Mixture of Potassium dihydrogen phosphate buffer: Acetonitrile 3 210 nm		1 ml/min	27	
21	C18 (250 x 4.6 mm, 5µm)	Mixture of Acetonitrile: Water (60:40 % v/v)		272 nm	1ml/min	28

Table II: Summery of methods related to HPLC technique

Table III: Summery of methods related to HPLC technique with plasma

S. No	Stationary Phase (Column)	nary Phase (Column) Mobile Phase (with ratio)			Flow rate	Reference		
	Sitagliptin AS Single Formulation							
1	C18 (150 x 4.6 mm, 5 µm)	Mixture of acetonitrile: methanol: buffer (2:3:5 % v/v/v)	4	267 nm	1ml/min	29		

Table IV: Summery of Analysis of Sitagliptin by UV-Spectroscopy methods.

S. No	Drug	Method	Description	Reference
1	Estimation Sitagliptine Phosphate Monohydrate in Pure and Tablet Dosage Form	Spectroscopic Method	Detection wavelength: 267 nm in Methanol Linearity range: 20-60 μg/ml Co-relation Co-efficient: 0.991 % Recovery range: 99.62-100.48 % %RSD: ≤2%	30
2	Estimation Sitagliptine by derivative spectroscopy	UV- Spectroscopic Method First order and Second order derivatives	Detection wavelength: Zero order derivative λ max: 267 nm in Methanol First order derivative λ max: 213 nm in Methanol Second order derivative λ max: 276 nm in Methanol Linearity range: 20-60 µg/ml Co-relation Co-efficient: 0.991 % Recovery range: Near 100% %RSD: ≤2%	31
3	Estimation of Metformin amd Sitagliptin	UV- Spectroscopic Simultaneous equation Method	Detection wavelength: 266 nm for Sitagliptin & 232 nm for Metformin Linearity range: 20-80 µg/ml for Sitagliptin & 10-50 µg/ml for Metformin Co-relation Co-efficient: 0.9992 for Metformin & 0.9997 for Sitagliptin % Recovery range: 99.62- 100.48 % %RSD: ≤2%	32
4	Estimation Sitagliptine Phosphate	Absorption Ratio and Area under curve Method	Detection wavelength: 267 nm in Methanol Linearity range: 20-160 μg/ml Co-relation Co-efficient: 0.991 % Recovery range: 99.62-100.48 % %RSD: ≤2%	33

Quality by Design

For improving the analytical method presently Quality by Design technique is used widely. Quality by design (QBD) which is discussed in ICH Q8, ^[1] Q9 and Q2 is well established for the development and manufacture of pharmaceuticals ^[34].

Benefits of Quality by Design Method

It helps in the development of a robust method. As per design setup sources of variability can be better controlled. Method Transfer success is greater when a method is transferred from research level to quality control department. This technique gives a space for the invention of new techniques by continuous improvement throughout the lifecycle ^[35].

Conclusion

This review depicts the reported Spectrophotometric and Chromatographic methods; developed and validated for estimation of Sitagliptin. According to this review it was concluded that for Sitagliptin different Spectroscopic & Chromatographic methods are available for single component as well as for combination and also it was found that the mobile phase containing phosphate buffer, methanol and acetonitrile were common for most of the chromatographic method to provide more resolution. It was observed that most common combination of Sitagliptin was with Metformin (ex. JANUMET). For chromatographic method flow rate is observed in the range of 0.8-1.5 ml/min to get good retention time. For most of the Spectroscopic methods common solvent is Methanol. Hence this all methods found to be simple, accurate, economic, precise, and reproducible in nature. But from this review it was clear that available methods can be improved by using Design of Expert (DOE) technique, which will give more accurate and precise result.

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